

Rhodium-Catalyzed Asymmetric 1,4-Addition to 1-Alkenylphosphonates

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Optically active phosphonic acid derivatives are important compounds because of their synthetic utility as chiral building blocks¹ as well as their potential biological activity.² Asymmetric 1,4-addition of organometallic reagents to α,β -unsaturated compounds is a powerful tool for carbon-carbon bond formation with simultaneous introduction of a new stereogenic carbon center at the β -position. Although many papers have appeared on the topic of catalytic asymmetric 1,4-addition to α,β -unsaturated carbonyl compounds with high enantioselectivity,³ to our best knowledge the enantioselective reaction to α,β -unsaturated phosphonates has not been reported yet,^{4,5} probably due to their low reactivity toward the 1,4-addition. Recently, we found asymmetric 1,4-addition of aryl- and alkenylboronic acids to α,β -unsaturated ketones which proceeds with high enantioselectivity under catalysis by a chiral phosphine-rhodium complex.⁶ Here we report that the rhodium-catalyzed asymmetric 1,4-addition is successfully applied to α,β -unsaturated phosphonates⁷ by use of triarylcyclotriboroxanes as arylating reagents in place of arylboronic acids.

We prepared geometrically pure diethyl (*E*)- and (*Z*)-1-propenylphosphonates (**1a**) by the palladium-catalyzed cross-coupling type reaction⁸ of diethyl phosphite with (*E*)- and (*Z*)-1-propenyl bromide, respectively. Treatment of the α,β -unsaturated phosphonate with phenylboronic acid under the conditions previously reported⁶ for α,β -unsaturated ketones gave a poor yield of diethyl 2-phenylpropylphosphonate (**3am**, Scheme 1). For example, the reaction of (*E*)-**1a** with phenylboronic acid in the presence of 3 mol % of the catalyst generated from Rh(acac)-(C₂H₄)₂ and (*S*)-binap in dioxane/H₂O (10/1) at 100 °C for 5 h gave **3am** (84% ee) only in 44% yield (entry 1 in Table 1). It was found that the rhodium catalyst loses its catalytic activity within 30 min under the conditions described above and that the presence of a large amount of water as a cosolvent causes the

(1) For a review on the use of phosphonates for alkene synthesis, see: Kelly, S. E. In *Comprehensive Organic Synthesis*; Trost, B. M., Fleming, I., Eds.; Pergamon: Oxford, 1991; Vol. 1, Chapter 3.1.

(2) For examples of asymmetric synthesis of biologically active phosphonates, see: (a) Blazis, V. J.; Koeller, K. J.; Spilling, C. D. *J. Org. Chem.* **1995**, *60*, 931. (b) Arai, T.; Bougauchi, M.; Sasai, H.; Shibasaki, M. *J. Org. Chem.* **1996**, *61*, 2926 and references therein. (c) Nagaoka, Y.; Tomioka, K. *J. Org. Chem.* **1998**, *63*, 6428.

(3) For reviews, see: (a) Schmalz, H.-G. In *Comprehensive Organic Synthesis*; Trost, B. M., Fleming, I., Eds.; Pergamon: Oxford, 1991; Vol. 4, Chapter 1.5. (b) Rossiter, B. E.; Swingle, N. M. *Chem. Rev.* **1992**, *92*, 771. (c) Noyori, R. *Asymmetric Catalysis in Organic Synthesis*; John Wiley and Sons: New York, 1994; pp 207–212. (d) Nógrádi, M. *Stereoselective Synthesis*; VCH Publishers: New York, 1995; pp 213–224. (e) Seyden-Penne, J. *Chiral Auxiliaries and Ligands in Asymmetric Synthesis*; John Wiley and Sons: New York, 1995.

(4) For a review on 1,4-addition reactions, see: Perlmutter, P. *Conjugate Addition Reactions in Organic Synthesis*; Pergamon Press: Oxford, 1992.

(5) For an example of nonasymmetric addition of alkyl- and vinylcopper reagents to α,β -unsaturated phosphonates, see: Nicotra, F.; Panza, L.; Russo, G. *J. Chem. Soc., Chem. Commun.* **1984**, 5.

(6) (a) Takaya, Y.; Ogasawara, M.; Hayashi, T.; Sakai, M.; Miyaura, N. *J. Am. Chem. Soc.* **1998**, *120*, 5579. (b) Takaya, Y.; Ogasawara, M.; Hayashi, T. *Tetrahedron Lett.* **1998**, *39*, 8479.

(7) For a review on vinylphosphonates in organic synthesis, see: Minami, T.; Motoyoshiya, J. *Synthesis* **1992**, 333.

(8) Hirao, T.; Masunaga, T.; Yamada, N.; Ohshiro, Y.; Agawa, T. *Bull. Chem. Soc. Jpn.* **1982**, *55*, 909.

Scheme 1

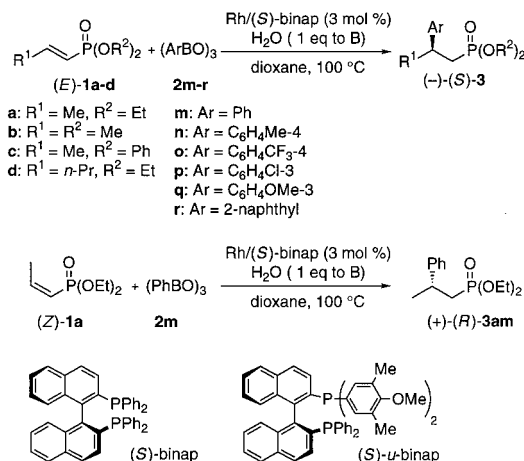


Table 1. Asymmetric 1,4-Addition of Arylboroxines **2** to 1-Alkenylphosphonates **1** Catalyzed by (*S*)-binap-Rhodium(I)^a

entry	phosphonate 1	(ArBO) ₃ 2	yield ^b (%) of 3	% ee ^c	[α] _D ²⁰ (c in CHCl ₃)
1 ^d	(<i>E</i>)- 1a	PhB(OH) ₂	44 (3am)	84 (<i>S</i>)	
2	(<i>E</i>)- 1a	2m	94 (3am)	96 (<i>S</i>)	-19 (0.93)
3 ^e	(<i>E</i>)- 1a	2m	5 (3am)		
4 ^d	(<i>E</i>)- 1a	<i>p</i> -TolB(OH) ₂	43 (3am)	86	
5	(<i>E</i>)- 1a	2n	84 (3an)	95	-23 (0.98)
6 ^f	(<i>E</i>)- 1a	2n	88 (3an)	96	
7	(<i>E</i>)- 1a	2o	64 (3ao)	96	-25 (0.86)
8	(<i>E</i>)- 1a	2p	61 (3ap)	96	-24 (0.72)
9	(<i>E</i>)- 1a	2q	81 (3aq)	95	-21 (0.92)
10	(<i>E</i>)- 1a	2r	89 (3ar)	89	-21 (1.01)
11 ^f	(<i>E</i>)- 1a	2r	92 (3ar)	90	
12	(<i>E</i>)- 1b	2m	96 (3bm)	94	-25 (1.07)
13	(<i>E</i>)- 1c	2m	95 (3cm)	91 (<i>S</i>)	-16 (1.10)
14 ^f	(<i>E</i>)- 1c	2m	99 (3cm)	94 (<i>S</i>)	
15	(<i>E</i>)- 1d	2m	39 (3dm)	99	-10 (1.03)
16	(<i>Z</i>)- 1a	2m	96 (3am)	89 (<i>R</i>)	+18 (1.13)
17 ^g	(<i>Z</i>)- 1a	2m	23 (3am)	97 (<i>R</i>)	
18 ^f	(<i>Z</i>)- 1a	2m	98 (3am)	92 (<i>R</i>)	

^a The reaction was carried out with phosphonate **1** (0.20 mmol), arylboroxine **2** (0.67 mmol), and H₂O (2.0 mmol) in dioxane (0.8–1.0 mL) at 100 °C for 3 h in the presence of 3 mol % of the catalyst generated from Rh(acac)(C₂H₄)₂ and (*S*)-binap unless otherwise noted.

^b Isolated yield by silica gel chromatography. ^c Determined by HPLC analysis with chiral stationary phase columns: Daicel Chiralcel AD (**3am**, **3an**, **3ao**, **3ap**, **3aq**, **3ar**, **3bm**) (eluent, hexane/2-propanol = 98/2), OD-H (**3cm**) (eluent, hexane/2-propanol = 90/10), and OJ (**3dm**) (eluent, hexane/2-propanol = 98/2). ^d Reaction of ArB(OH)₂ in dioxane/H₂O (10/1). ^e Reaction without addition of H₂O. ^f As a chiral ligand, (*S*)-*u*-binap was used in place of (*S*)-binap. ^g Reaction was stopped at the reaction period of 3 min.

catalyst deactivation. The asymmetric 1,4-addition was greatly improved by carrying out the reaction with triphenylcyclotriboroxane (phenylboroxine, (PhBO)₃)⁹ (**2m**) in place of phenylboronic acid (entry 2). Thus, the reaction of (*E*)-**1a** with phenylboroxine (**2m**) and 1 equiv (to boron) of water in dioxane at 100 °C for 3 h gave 94% yield of **3am** ([α]_D²⁰ -19 (c 0.93, chloroform)), whose enantiomeric purity was determined to be 96% ee by HPLC analysis with a chiral stationary phase column (entry 2). The absolute configuration of (*-*)-**3am** was assigned to be *S* by correlation with (+)-(*R*)-1,3-diphenyl-1-butene (**4**)¹⁰

(9) Arylboroxines are readily obtained by dehydration of arylboronic acids by azeotropic removal of water from their xylene solution or heating at 300 °C in vacuo. For a pertinent review, see: Lappert, M. F. *Chem. Rev.* **1956**, *56*, 959.

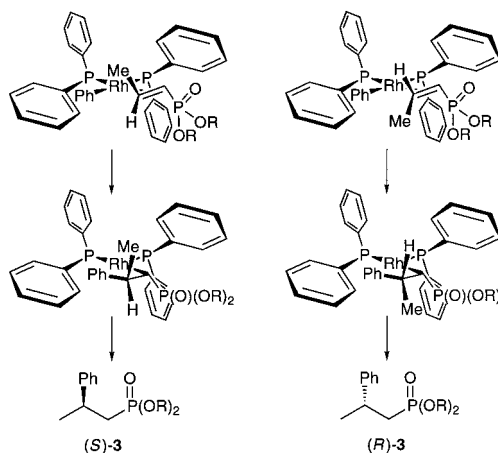
(vide infra). The addition of 1 equiv of water is essential for the high yield,¹¹ almost no reaction taking place in the absence of water (entry 3).

Under similar reaction conditions, diethyl (*E*)-1-propenylphosphonate ((*E*)-**1a**) underwent asymmetric arylation with some other arylboroxines (**2n–2r**) to give the corresponding diethyl 2-arylpropylphosphonates (**3an–3ar**) in good yields with high enantioselectivity (entries 5, and 7–10). Here, again, the yield of **3an** was much lower in the reaction with *p*-tolylboronic acid than with *p*-tolylboroxine (entries 4 and 5). The asymmetric phenylation was also successful for dimethyl and diphenyl esters of (*E*)-1-propenylphosphonate ((*E*)-**1b,c**) (entries 12 and 13). The enantioselectivities and chemical yields were slightly higher in the reaction catalyzed by rhodium complex coordinated with unsymmetrically substituted binap ligand, (*S*)-*u*-binap,¹² which has diphenylphosphino and bis(3,5-dimethyl-4-methoxyphenyl)phosphino groups at the 2 and 2' positions on the 1,1'-binaphthyl skeleton (entries 6, 11, and 14).

The rhodium-catalyzed asymmetric phenylation of the *Z* isomer of diethyl 1-propenylphosphonate (*Z*)-**1a** with phenylboroxine **2m** for 3 h gave the *R* isomer of **3am** with 89% ee (entry 16). The observation of the opposite absolute configuration of **3am** for (*E*)-**1a** and (*Z*)-**1a** indicates that the dialkoxyphosphinyl moiety on the 1-alkenylphosphonate plays a key role in the enantioface selection (Scheme 2). The (*S*)-binap-rhodium catalyst recognizes the enantioface of 1-propenylphosphonate by the steric bulkiness of the phosphinyl group; both (*E*)-**1a** and (*Z*)-**1a** phenylated on the rhodium from the *1si* face irrespective of the *E,Z* geometry of the 1-propenyl moiety.¹³ The *Z* isomer (*Z*)-**1a** was found to undergo slow isomerization into the *E* isomer under these reactions conditions, resulting in the loss of enantioselectivity. This was demonstrated by stopping the reaction at the reaction period of 3 min, which gave **3am** of 97% ee (entry 17).

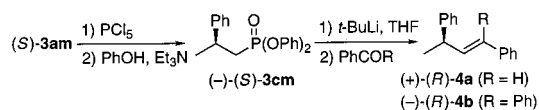
The optically active alkylphosphonates **3**, containing the stereogenic carbon center at the β -position, can be used as chiral building blocks for the synthesis of optically active alkenes by the Horner–Emmons type reaction (Scheme 3). The olefination of carbonyl compounds with diphenyl phosphonates proceeded without loss of enantiomeric purity. Unfortunately, the elimination forming alkenes did not take place from the β -hydroxyphospho-

Scheme 2^a



^a The binaphthylene moiety in (*S*)-binap is omitted for clarity.

Scheme 3



nate intermediates in the reaction of diethyl phosphonates,¹⁴ but diethyl esters were readily converted into diphenyl esters by way of dichlorides.¹⁵ For example, the ester substitution of (*−*)-**3am** (91% ee) from ethyl to phenyl followed by treatment of the resulting diphenyl ester (*−*)-**3cm** with *tert*-butyllithium and benzaldehyde gave (+)-(*R*)-(*E*)-1,3-diphenyl-1-butene (**4a**)¹⁰ of 92% ee,¹⁶ together with a minor amount of (*Z*)-isomer (*E/Z* = 82/18), indicating that the absolute configuration of **3am** and **3cm** is (*−*)-(*S*). Similarly, the reaction of (*−*)-**3cm** with benzophenone gave (*−*)-(*R*)-1,1,3-triphenyl-1-butene (**4b**)¹⁷ of 91% ee.¹⁶

To summarize, we have realized, for the first time, the catalytic asymmetric 1,4-addition to 1-alkenylphosphonates, forming 2-arylalkylphosphonates in high yields with high enantioselectivity, by use of a new catalytic system consisting of a chiral phosphine–rhodium catalyst and arylboroxines as arylating reagents.

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Supporting Information Available: Experimental procedures and spectroscopic and analytical data for the substrates and products (PDF). This material is available free of charge via the Internet at <http://pubs.acs.org>.

JA993184U

(14) Corey, E. J.; Kwiatkowski, G. T. *J. Am. Chem. Soc.* **1966**, *88*, 5654.
(15) Ando, K. *Tetrahedron Lett.* **1995**, *36*, 4105.

(16) The enantiomeric purities of (*E*)-**4a** and **4b** were determined by GLC analysis with CP-Chirasil-dex CB (25 m).

(17) [α]_D²⁰ −73 (*c* 0.52, chloroform). The absolute configuration was assigned by the correlation with the starting (*−*)-(*S*)-**3cm**.

(10) Hayashi, T.; Konishi, M.; Fukushima, M.; Mise, T.; Kagotani, M.; Tajika, M.; Kumada, M. *J. Am. Chem. Soc.* **1982**, *104*, 180.

(11) Interestingly, the rhodium-catalyzed reaction of (*E*)-**1a** with phenylboronic acid in nonaqueous dioxane, namely, in the absence of water, resulted in a lower yield (67%) of **3am** (88% ee), though a boronic acid is known to be in equilibrium with a boroxine and water (ref 9). Methanol can also be used in place of water as a protic additive to phenylboroxine (85% yield of **3am**), while the reaction with dimethyl ester (PhB(OMe)₂) did not take place (<2% yield).

(12) The new chiral bisphosphine (*S*)-*u*-binap was prepared starting from (*S*)-binaphthol ditriflate by a sequence of reactions consisting of palladium-catalyzed monophosphinylation with bis[(3,5-dimethyl-4-methoxy)phenyl]phosphine oxide, reduction of the phosphine oxide with trichlorosilane and triethylamine, and nickel-catalyzed cross-coupling of the remaining triflate with diphenylphosphine in 74% overall yield: [α]_D²⁰ −103 (*c* 1.08, chloroform).

(13) For the highly skewed structure of transition metal complexes coordinated with a binap ligand, see: Ozawa, F.; Kubo, A.; Matsumoto, Y.; Hayashi, T. *Organometallics* **1993**, *12*, 4188 and references therein.